Effects of different salts and mannitol on seed imbibition, germination and ion content of *Crithmum maritimum* L. (Apiaceae)

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Saline habitats contain various salt types that differently affect seed germination and subsequent seedling establishment. Most studies on the germination process have been carried out using NaCl, which may limit the accuracy of the conclusions drawn, especially regarding the plant behaviour under field conditions. We compare here the impact of iso-osmotic solutions of four salts (NaCl, Na₂SO₄, MgCl₂ and MgSO₄) and mannitol on seed imbibition, germination, germination recovery, viability and seedling ion relations of the halophyte *Crithmum maritimum* L. Seed imbibition percentage declined as osmotic potentials decreased. MgCl₂ and NaCl were the most and less adverse salt on the germination, respectively. As for seed germination and viability, K⁺, Ca²⁺, NO₃⁻ and PO₄³⁻ contents within seedlings appeared to be salt-diminished. This was associated with higher ion (K⁺ and Ca²⁺) leakage. In conclusion, both osmotic and toxic salt components impaired seed germination. This may be related to the effects of different salts on ion content, seed imbibition, and nutrient leakage through membranes.

Key words: Crithmum maritimum L., halophytes, ion relations, salinity, seed germination.

INTRODUCTION

Seed germination starts with water uptake by the dry seed, i.e. imbibition, and ends with radicle elongation (Bewley & Black, 1994). Water uptake under favourable conditions, after the breaking down of seed dormancy if present, leads to the activation of metabolic processes, so that germination can successfully take place. In halophytes, which undergo substantial fluctuation in soil salinity levels over time and space, imbibition is the most affected procedure by salt exposure process (Ungar, 1996). This may partly explain why these species (native of saline habitats) may behave as salt-sensitive species at the germinative stage.

Salt presence in the medium frequently impairs seed germination (Ungar, 1996), either by restricting water supply (osmotic effect), or by causing a specific injury related to ion toxicity (ionic effect) (Duan et al., 2004). Seed germination in salt-affected soils may be influenced not only by the total concentration of dissolved salts, but also by the salt type (Ryan et al., 1975). Indeed, it is well known that saline soils contain multiple salt constituents that have different physiological effects on plant development (Tobe et al., 2002, 2004). The most common salt components in saline soils are Na⁺, Mg²⁺, and Ca²⁺ cations, and Cl⁻ and SO₄²⁻ anions (Shainberg, 1975). However, most of the studies dealing with the salinity effects on seed germination have been focusing on the effect of a single salt, namely NaCl, and explained the germination inhibition by the osmotic component of salts, *i.e.* the restriction of water uptake (Egan et al., 1997; Keiffer & Ungar, 1997; Gul & Weber, 1998; Pujol et al., 2000; Debez et al., 2004). Therefore, little is known about the impact of other salts on germination (Sosa et al., 2005). Further, data regarding seed imbibition under salt conditions for halophytes are scarce.

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Crithmum maritimum L. is a typical halophyte of rocky coastal ecosystems, which is subjected to regular variation in their climatic conditions (Abdelly et al., 2006). This species shows substantial economical and medicinal importance: its leaves display high antioxidant and antimicrobial activities (Meot-Duros et al., 2008; Meot-Duros & Magné, 2009), whereas seeds are rich in essential oils and other bioactive substances (Atia et al., 2009). In addition, seeds contain significant amounts of oil, potentially edible due to its fatty acid composition close to olive oil (Zarrouk et al., 2003). In a preliminary work, we observed that salinities exceeding 50 mM NaCl, severely inhibited the germination of Tunisian population of C. maritimum L. (Atia et al., 2006). Hence, one may hypothesize that the combination of both osmotic and ionic components of salt may inhibit seed germination of C. maritimum L. As for halophytes, seed germination of C. maritimum L. may be strongly influenced by the nature of ions, such as Na+, Mg2+, Ca2+, Cl-, and SO_4^{2-} . We address in this study the impact of iso-osmotic solutions of different salts (NaCl, Na₂SO₄, MgCl₂, MgSO₄) and mannitol on: i) seed germination capacity and imbibition over time, ii) seedling ion accumulation, and iii) seedling cell membrane stability (assessed by Ca^{2+} and K^+ leakage).

MATERIALS AND METHODS

Fruit harvesting and seed germination experiments

Mature fruits were collected in December 2005 from Tabarka (NW of Tunisia) and were stored dry under laboratory conditions at 18-23°C. The fruit of C. maritimum L. is a schizocarp divided into two mericarps. The mericarp is composed of a spongy outer coat, a secretory envelope delimiting the large endosperm, and a rudimentary embryo (Fig. 1). At fruit maturation, the mesocarp is a spongy coat that is easy to remove. However, the endocarp and the secretory envelope remain firmly attached to the seed. Therefore, in this study the term seed refers to the seed and the endocarp and the secretory envelope. In May 2006, after the spongy coat was removed, seeds were disinfected in calcium hypochlorite solution for 5 min before starting the germination tests. Twenty five seeds were sown in 9-cm diameter Petri dishes and covered with a double layer of filter paper (type *Filtrak*), moistened with 5 ml distilled water or a suitable solution containing either NaCl, Na₂SO₄, MgCl₂, MgSO₄, or mannitol (Sigma Chemicals) in concentrations calculated to obtain the following osmotic potentials:



FIG. 1. Stereomicroscope photograph of *C. maritimum* L. fruit, showing the seed within the spongy coat.

-0.4, -0.8, and -1.2 MPa (Tobe *et al.*, 2002; Sosa *et al.*, 2005). The Petri dishes were sealed with transparent plastic film to prevent solution evaporation. Every week, the solution and the filter paper were changed. No changes in the solution volume in the Petri dishes were observed during the treatment. Seed germination was carried out in growth chambers under the following conditions: 16/8 hrs and 23°/18°C temperature and light/dark regime. White light was produced by five fluorescent lamps (Type OSRAM 40 W, fluorescence rate of 25 µmol m⁻² s⁻¹, 400-700 nm).

Germination and imbibition parameters

Seed imbibition percentage was determined after 8 days as the relative water content (RWC) using the following formula: RWC (%) = $[(FW - DW)/DW] \times 100$, where DW and FW stand for seed initial dry weight and seed fresh weight, respectively (Li *et al.*, 2005).

Germinated seeds were counted at two days interval, a seed being considered as germinated at the radicle appearance. Besides the final germination percentage, the index of germination velocity was determined according to a modified Timson's index (Khan & Ungar, 1984) as: S = G/T, with G representing the germination percentage, calculated at two days interval and T the total germination time. The index of germination recovery was calculated using the relation $[(a-b)/(c-b)] \times 100$, where a is the total number of seeds germinated after being transferred to distilled water, b is the total number of seeds germinated in saline solution, and c is the total number of seeds germinated in saline or mannitol solution plus those transferred on distilled water (previously non germinated in presence of salts or mannitol).

Ion relations

Ions were extracted by boiling grinded seeds previously germinated at -0.4 MPa at 100° C for 10 min. K⁺ and Na⁺ were assayed by flame photometry, Ca²⁺ and Mg²⁺ by atomic absorption spectrophotometry, and Cl⁻, NO₃⁻, SO₄²⁻, and PO₄³⁻ by ion chromatography. In order to assess the salt treatment effect on membrane stability, 25 seeds were sown in 9-cm diameter Petri dishes without using filter paper, and moistened with 5 ml of the suitable solution (i.e. distilled water, saline or mannitol) at -0.4 MPa. Cation (K⁺ and Ca²⁺) leakage was assessed at 10 days after imbibition.

Statistical analysis

Data presented are the mean values of 4 replicates $(\pm SE)$. A two-way ANOVA was used, using the SPSS

10.0 for Windows software, in order to assess the significance of the osmotic agent (OA), i.e. salt or mannitol, the osmotic potential (OP) and their interaction (OA*OP) on the seed imbibition and germination. Before the statistical analysis, germination data were arcsine transformed.

RESULTS

Salt effect on seed imbibition

The two-way ANOVA revealed a significant effect of osmotic agent (OA) and osmotic potential (OP) on the seed imbibition percentage after 8 days of treatment, but no significant effect of their interaction (OA × OP) was found (Table 1). The lower was the external osmotic potential, the lower seed imbibition (Fig. 2). The mannitol-treated seeds showed a higher imbibition percentage as compared to those treated with salt. For instance, the imbibition capacity was not significantly affected by mannitol at -0.4 MPa and remained higher than in the salt-treated seeds at -0.8 MPa and -1.2 MPa, respectively (Fig. 2). Na₂SO₄ was the most adverse salt on seed imbibition, NaCl was the less one, whereas MgSO₄ and MgCl₂ impacts were intermediate (Fig. 2).

Salt effect on seed germination and seed viability

The two-way ANOVA showed that OA, OP and their interaction had a significant impact on the final germination percentage and germination velocity (p < 0.001)



FIG. 2. Effect of 8 day-long exposure with monosaline solutions (osmotic potential: -0.4, -0.8, and -1.2 MPa) of NaCl, Na₂SO₄, MgCl₂, MgSO₄ and mannitol on imbibition percentage (mean ± SE).

TABLE 1. F-values of two-way ANOVA analysis for characteristics of seed germination and imbibition for se	eds moistened
with distilled water, different salts (NaCl, Na2SO4, MgCl2, and MgSO4) and mannitol at -0.4, -0.8 and -1.2 M	Pa

Independent variable	Osmotic potential (OP)	Osmotic agent (OA)	$OA \times OP$	
Imbibition percentage	51.639****	5.268***	0.479 ^{NS}	
Final germination percentage	398.493****	75.433****	19.622****	
Germination velocity	627.076****	71.481****	17.647****	
Seed viability	7.806****	4.351**	1.920 ^{NS}	
Recovery	9.584****	2.951*	1.204 ^{NS}	

*significant difference at p < 0.05, **significant difference at p < 0.01, ***significant difference at p < 0.001, ***significant difference at p < 0.001, NS no significant difference



FIG. 3. Final germination percentage (bars) and germination velocity (curve) for seeds germinated in saline and mannitol solutions (mean \pm SE).

(Table 1). Mannitol treatment up to -0.8 MPa did not affect final seed germination percentage, but delayed the germination process, since germination velocity was reduced (Fig. 3A). At -0.4 MPa, seed germination was significantly affected and delayed by Na₂SO₄ and MgCl₂ (Fig. 3C and D), whereas NaCl and MgSO₄ reduced only seed germination velocity (Fig. 3B and E). Lowering of the osmotic potentials, suppressed the germination process, irrespective of salt treatment (Fig. 3B, C, D and E). Both seed germination recovery and seed viability were significantly influenced by OA and OP but not their interaction (Table 1). While mannitol strongly altered seed viability at -1.2 MPa (Fig. 4A), a specific salt effect was observed for both germination recovery and seed viability, especially at -1.2 MPa, in the following decreasing order: MgCl₂, MgSO₄, NaCl, and Na₂SO₄ (Fig. 4D, E, B and C).

Salt effect on seedling ion content

There was a high significant effect (p < 0.0001) of OA and OA×OP on the seedling Na⁺ content (Table 2), the highest accumulation being recorded for NaCl and Na₂SO₄ treatments (Table 2). OA, OP and their interaction also significantly affected (p < 0.0001) K⁺ content (Table 2). The treatment with sodium salts (NaCl and Na₂SO₄) reduced seedling K⁺ content as compared to the control. In contrast, the MgSO₄treated seeds showed the highest K⁺ content (Table



FIG. 4. Germination recovery (full bars) and seed viability (hatched bars) for seeds germinated in saline and mannitol solutions (mean ± SE).

Independent variable	Osmotic potential (OP)	Osmotic agent (OA)	OA × OP	
Na ⁺	3.810 ^{NS}	102.284****	101.005****	
K^+	250.898****	97.045****	86.322****	
Ca ²⁺	0.769 ^{NS}	4.514**	5.071**	
Mg^{2+}	202.430****	29.933****	29.273****	
Cl-	8.603****	27.164****	22.894****	
NO_3^-	24.895****	1.019 ^{NS}	1.260 ^{NS}	
PO 4 ³⁻	15.262****	31.964****	30.802****	
SO ₄ ^{2–}	326.941****	174.905****	174.762****	

TABLE 2. F-values of two-way ANOVA analysis of seed ion relations. Seeds were moistened with either distilled water or iso-osmotic solutions (-0.4 MPa) of salts (NaCl, Na₂SO₄, MgCl₂, and MgSO₄) and mannitol

*significant difference at p < 0.05, **significant difference at p < 0.01, ***significant difference at p < 0.001, ***significant difference at p < 0.001, NS no significant difference



FIG. 5. Effect of monosaline solutions (osmotic potential: -0.4 MPa) of NaCl, Na₂SO₄, MgCl₂, MgSO₄ and mannitol on calcium and potassium leakage (mg seed⁻¹). Both parameters were calculated after 10 days from the beginning of imbibition (mean ± SE).

potential solutions (mean \pm SE)						
	H ₂ O	Mannitol	NaCl	Na ₂ SO ₄	MgCl ₂	MgSO ₄
Na ⁺	2.50 ± 0.48	2.33 ± 0.16	6.5 ± 0.28	6.50 ± 0.48	2.75 ± 3.20	2.66 ± 0.16
K ⁺	5.26 ± 0.68	5.66 ± 0.73	2.6 ± 0.11	4.36 ± 0.42	6.53 ± 0.36	7.23 ± 0.72
Ca ²⁺	1.70 ± 0.21	1.99 ± 0.22	1.5 ± 0.36	2.02 ± 0.01	1.09 ± 0.12	1.81 ± 0.08
Mg ²⁺	0.78 ± 0.07	0.87 ± 0.10	1.18 ± 0.12	0.84 ± 0.15	1.59 ± 0.18	1.46 ± 0.08
Cl-	3.88 ± 0.88	3.05 ± 0.82	6.60 ± 0.42	3.01 ± 1.27	7.01 ± 0.64	2.60 ± 0.75
NO_3^{-}	0.19 ± 0.05	0.16 ± 0.07	0.14 ± 0.00	0.11 ± 0.00	0.10 ± 0.03	0.15 ± 0.03
PO 4 ^{3–}	5.49 ± 1.30	6.64 ± 0.73	7.85 ± 0.74	1.69 ± 0.16	1.16 ± 0.51	5.57 ± 0.91
SO ₄ ²⁻	1.10 ± 0.12	0.45 ± 0.13	1.53 ± 0.44	7.11 ± 1.24	0.36 ± 0.10	6.16 ± 0.70

TABLE 3. Seedling ion content (mg g⁻¹ dry weight) after 10 days of imbibition in either distilled water or in -0.4 MPa osmotic potential solutions (mean \pm SE)

3). OA and OA×OP affected significantly the Ca²⁺ (p < 0.01), Mg²⁺, Cl⁻, PO₄³⁻ and SO₄²⁻ (p < 0.0001) contents (Tables 2 and 3), but only OP had a significant impact on the NO₃⁻ content (p < 0.0001) (Table 2). Sulphate salts (MgSO₄ and Na₂SO₄) decreased Cl⁻ contents in seedlings as compared to Cl⁻ salts (Table 3). Moreover, the seeds exposed to Na₂SO₄ and MgCl₂ showed lower NO₃⁻ and PO₄³⁻ contents (Table 3). OA and OA×OP significantly affected the SO₄⁻ content (p < 0.0001) (Table 2). MgCl₂ treatment reduced the SO₄²⁻ content. In contrast, NaCl treatment increased SO₄²⁻ content in comparison to the control (Table 3).

Salt effect on membrane stability

At -0.4 MPa, the chloride salts (NaCl and MgCl₂) as well as mannitol led to higher calcium leakage as compared to the control, whereas sulphate salts had no significant effect (Fig. 5). Seed treatment with sodium salts (NaCl and Na₂SO₄) caused higher potassium release in the external medium, while no significant trend was observed for mannitol and the remaining salts (Fig. 5).

DISCUSSION

Whether osmotic (Gul & Weber, 1998), toxic (Al-Karaki, 2001; Tobe et al., 2002) or combination of both (Rehman et al., 1996; Katembe et al., 1998; Sosa et al., 2005), the nature of salt effect on seed germination in halophytes is still controversial, despite being intensively investigated. Low water potentials reduce water uptake by seeds, thereby inhibiting germination (Katembe et al., 1998). Our results indicate that the different salt solutions as well as mannitol solution decreased water uptake (in terms of imbibition percentage and velocity), final germination percentage, and germination velocity in C. maritimum L. Consistent with previous reports on Prosopis strombulifera (Sosa et al., 2005), Atriplex patula and A. prostrata (Katembe et al., 1998), we found that imbibition, and subsequent germination were less affected by mannitol than by other salts. Both germination recovery and seed viability were particularly altered at the lowest osmotic potential (-1.2 MPa). Thus, salinity may inhibit seed germination of C. maritimum L. through the combination of both osmotic and ionic components. Crithmum maritimum L. seed germination was not only affected by salt concentration, but also by the nature of the salt applied. This is in agreement

with previous studies on halophytic species like *Puccinellia festucaeformis* (Ungar, 1996), *Chenopodium glaucum* (Duan *et al.*, 2004), *Prosopis strombulifera* (Sosa *et al.*, 2005), *Sporobolus madraspatanus*, and *Aeluropus lagopoides* (Joshi *et al.*, 2005).

The strongest inhibition on the final germination percentage, germination recovery, and seed viability was observed with MgCl₂, while MgSO₄ altered seed viability only at the lowest potential (-1.2 MPa). Na₂SO₄ and NaCl had not effect on seed viability, despite the strong inhibition of germination observed at -0.8 and -1.2 MPa. The intensity at which salts impair seed germination greatly varies among halophytes. For instance, the order was MgCl₂ > NaCl > KCl > MgSO₄ in *Puccinellia festucaeformis* seeds (Katembe *et al.*, 1998), MgCl₂ > Na₂SO₄ > NaCO₃ > NaCl > MgSO₄ in *Chenopodium glaucum* (Duan *et al.*, 2004), and KCl > NaCl > MgSO₄ > MgCl₂ > Na₂SO₄ (Joshi *et al.*, 2005) in *Sporobolus madraspatanus* and *Aeluropus lagopoides*.

The analysis of ion relations indicated that Na+treated seeds showed similar contents of Ca²⁺ and Mg²⁺ to those of the control. However, they were characterised by lower K⁺ content in concomitance with higher K⁺ leakage, which may suggest that the Na⁺-salt specific effect was related to potassium loss. Such a pattern was also reported in Haloxylon ammodendron (Tobe et al., 2004) and in several species thriving in Mediterranean salt marshes (Alvarez-Rogel et al., 2006). These authors observed decreasing contents of Ca²⁺ and K⁺ as salinity increased, leading to an imbalance in favor of the most toxic cations, such as Na⁺ and Mg²⁺. K⁺ is an important osmoticum and high K⁺/Na⁺ ratio is often considered as indicative of salt tolerance (Muhammed et al., 1987; Hosseini et al., 2002). The regulation of Na⁺ and K⁺ uptake and their distribution in plant tissues is notably ensured by highly K+-selective inward- and outward-rectifying K⁺ channels (Maathuis & Amtmann, 1999) and Na⁺/H⁺ antiporters, that extrude out Na⁺ against a concentration gradient (Blumwald et al., 2000). As stated by Akram et al. (2009), lower mineral uptake under saline conditions may be ascribed to Na⁺-induced blockage or reduced activity of these transporters.

NaCl and Na_2SO_4 treated seeds displayed a high germination recovery percentage, which indicates that the salt specific effects are rather osmotic. In pepper, Na_2SO_4 as well as $MgCl_2$ treated seeds, showed a strong decline of P content, in concomitance with the loss of membrane integrity (Xu & Kafkafi, 2003). The two above-cited salts decreased the phosphate and nitrate contents in our experiment, which could also explain the seed lower germination capacity. Sulphate content was also highly affected in MgCl₂ treated seeds. In plants, sulphate is a major component of proteins, especially of the amino acids cysteine and methionine and the vitamin thiamin and the coenzyme A (Epstein, 1972). Sulphate loss from seeds is considered as an indication of amino acid and vitamin degradation (Ouyang et al., 2002), hence highlighting the toxic effect of MgCl₂. It is noteworthy that in our conditions, MgCl₂-exposed seeds showed the lowest germination capacity and seed viability. This was associated with the lowest Ca²⁺ content and the highest Ca²⁺ leakage. Calcium is essential to preserve the structural and functional integrity of plant membranes, stabilize cell wall structures, regulate ion transport and selectivity, and control ion-exchange behavior as well as cell wall enzyme activities (Rengel, 1992; Marschner, 1995). These functions may be seriously impaired under low Ca²⁺ availability, due to calcium displacement from its membrane binding sites by other cations. For instance, calcium deficiency symptoms are common when Na⁺/Ca²⁺ ratio is high in soil water (Shannon & Grieve, 1999; Tobe et al., 2004).

In conclusion, both osmotic and toxic salt components impaired seed germination of C. maritimum L. by reducing seed imbibition and ion content, and increasing nutrient leakage. The salt-induced inhibition of germination was salt-specific and in following order: $MgCl_2 > MgSO_4 > Na_2SO_4 > NaCl. MgCl_2 ex$ erted a strong toxic effect that may be explained by the highest loss of nutrients, especially phosphorus, nitrate, sulphate, and calcium. MgSO4 decreased germination by a toxic effect only at the lowest osmotic potential (-1.2 MPa), whereas Na₂SO₄ and NaCl inhibited germination mainly via an osmotic effect, since high germination recovery was registered after seed transfer in distilled water. Both Na⁺-containing salts caused a high leakage of K⁺. Crithmum maritimum L. usually grows on rocky coasts in the vicinity of seawater, which contains multiple ions among which Na⁺, Mg²⁺, Ca²⁺, Cl⁻, and SO₄²⁻. The seasonal exposure to high salinity levels may reduce the availability of seed reserves, thereby impairing germination. Yet, seeds may remain viable and germinate after the winter rains, so that the plant can successfully establish.

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